

ENVIRONMENTAL CONTAMINANTS ENCYCLOPEDIA

COAL TAR ENTRY

July 1, 1997

COMPILERS/EDITORS:

ROY J. IRWIN, NATIONAL PARK SERVICE

WITH ASSISTANCE FROM COLORADO STATE UNIVERSITY

STUDENT ASSISTANT CONTAMINANTS SPECIALISTS:

MARK VAN MOUWERIK

LYNETTE STEVENS

MARION DUBLER SEESE

WENDY BASHAM

NATIONAL PARK SERVICE

WATER RESOURCES DIVISIONS, WATER OPERATIONS BRANCH

1201 Oakridge Drive, Suite 250

FORT COLLINS, COLORADO 80525

**WARNING/DISCLAIMERS:**

Where specific products, books, or laboratories are mentioned, no official U.S. government endorsement is implied.

Digital format users: No software was independently developed for this project. Technical questions related to software should be directed to the manufacturer of whatever software is being used to read the files. Adobe Acrobat PDF files are supplied to allow use of this product with a wide variety of software and hardware (DOS, Windows, MAC, and UNIX).

This document was put together by human beings, mostly by compiling or summarizing what other human beings have written. Therefore, it most likely contains some mistakes and/or potential misinterpretations and should be used primarily as a way to search quickly for basic information and information sources. It should not be viewed as an exhaustive, "last-word" source for critical applications (such as those requiring legally defensible information). For critical applications (such as litigation applications), it is best to use this document to find sources, and then to obtain the original documents and/or talk to the authors before depending too heavily on a particular piece of information.

Like a library or most large databases (such as EPA's national STORET water quality database), this document contains information of variable quality from very diverse sources. In compiling this document, mistakes were found in peer reviewed journal articles, as well as in databases with relatively elaborate quality control mechanisms [366,649,940]. A few of these were caught and marked with a "[sic]" notation, but undoubtedly others slipped through. The [sic] notation was inserted by the editors to indicate information or spelling that seemed wrong or misleading, but which was nevertheless cited verbatim rather than arbitrarily changing what the author said.

Most likely additional transcription errors and typos have been added in some of our efforts. Furthermore, with such complex subject matter, it is not always easy to determine what is correct and what is incorrect, especially with the "experts" often disagreeing. It is not uncommon in scientific research for two different researchers to come up with different results which lead them to different conclusions. In compiling the Encyclopedia, the editors did not try to resolve such conflicts, but rather simply reported it all.

It should be kept in mind that data comparability is a major problem in environmental toxicology since laboratory and field methods are constantly changing and since there are so many different "standard methods" published by EPA, other federal agencies, state agencies, and various private groups. What some laboratory and field investigators actually do for standard operating practice is often a unique combination of various standard protocols and impromptu "improvements." In fact, the interagency task force on water methods concluded that [1014]:

It is the exception rather than the rule that water-quality monitoring data from different programs or time periods can be compared on a scientifically sound basis, and that...

No nationally accepted standard definitions exist for water quality parameters. The different organizations may collect data using identical or standard methods, but identify them by different names, or use the same names for data collected by different methods [1014].

Differences in field and laboratory methods are also major issues related to (the lack of) data comparability from media other than water: soil, sediments, tissues, and air.

In spite of numerous problems and complexities, knowledge is often power in decisions related to chemical contamination. It is therefore often helpful to be aware of a broad universe of conflicting results or conflicting expert opinions rather than having a portion of this information arbitrarily censored by someone else. Frequently one wants to know of the existence of information, even if one later decides not to use it for a particular application. Many would like to see a high percentage of the information available and decide for themselves what to throw out, partly because they don't want to seem uninformed or be caught by surprise by potentially important information. They are in a better position if they can say: "I knew about that data, assessed it based on the following quality assurance criteria, and decided not to use it for this application." This is especially true for users near the end of long decision processes, such as hazardous site cleanups, lengthy ecological risk assessments, or complex natural resource damage assessments.

For some categories, the editors found no information and inserted the phrase "no information found." This does not necessarily mean that no information exists; it

simply means that during our efforts, the editors found none. For many topics, there is probably information "out there" that is not in the Encyclopedia. The more time that passes without encyclopedia updates (none are planned at the moment), the more true this statement will become. Still, the Encyclopedia is unique in that it contains broad ecotoxicology information from more sources than many other reference documents. No updates of this document are currently planned. However, it is hoped that most of the information in the encyclopedia will be useful for some time to come even with out updates, just as one can still find information in the 1972 EPA Blue Book [12] that does not seem well summarized anywhere else.

Although the editors of this document have done their best in the limited time available to insure accuracy of quotes as being "what the original author said," the proposed interagency funding of a bigger project with more elaborate peer review and quality control steps never materialized.

The bottom line: The editors hope users find this document useful, but don't expect or depend on perfection herein. Neither the U.S. Government nor the National Park Service make any claims that this document is free of mistakes.

The following is one chemical topic entry (one file among 118). Before utilizing this entry, the reader is strongly encouraged to read the README file (in this subdirectory) for an introduction, an explanation of how to use this document in general, an explanation of how to search for power key section headings, an explanation of the organization of each entry, an information quality discussion, a discussion of copyright issues, and a listing of other entries (other topics) covered.

See the separate file entitled REFERENC for the identity of numbered references in brackets.

HOW TO CITE THIS DOCUMENT: As mentioned above, for critical applications it is better to obtain and cite the original publication after first verifying various data quality assurance concerns. For more routine applications, this document may be cited as:

**Irwin, R.J., M. VanMouwerik, L. Stevens, M.D. Seese, and W. Basham.** 1997. Environmental Contaminants Encyclopedia. National Park Service, Water Resources Division, Fort Collins, Colorado. Distributed within the Federal Government as an Electronic Document (Projected public availability

on the internet or NTIS: 1998).

Coal Tar (CAS number 8007-45-2)

**Brief Introduction:**

**Br.Class:** General Introduction and Classification Information:

Coal tar is a viscous liquid mixture of hydrocarbon compounds, derived, along with coke, from the destructive distillation of coal in coking ovens [279]. According to the US Coast Guard Emergency Response Notification System (ERNS), coal tar is one of the most commonly spilled petroleum products in the U.S [635].

Coal tar itself may also be subjected to distillation, a process that separates groups of the components of coal tar from groups of others [751]. The coal tar is heated, causing a number of the components to vaporize at their boiling points; these vapors are then condensed to liquids [751]. The substances derived from this process are often called "coal tar distillates [751]." The US Department of Transportation (USDOT) regulates the transportation of coal tar distillates as flammable liquids [751].

Tars in general are obtained through destructive processing of coal, wood, or petroleum. Coal tar, asphalt, and various other petroleum products are all different products.

**Br.Haz:** General Hazard/Toxicity Summary:

The hazards of coal tar as a complex mixture are not well documented, but should relate to hazards from the constituents in coal tar. Coal tar consists of a mixture of many organic compounds, like benzene, toluene, phenol, naphthalene, anthracene, and others [751]. It is the aromatics that pose the most danger to both humans and fish and wildlife. Chronic effects of some of the constituents in coal tar include changes in the liver and harmful effects on the kidneys, heart, lungs, and nervous system [609,764,765,766,767]. For more information on hazards, see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

The aquatic toxicity in the water soluble fraction of a sample of middle distillate coal oil was mostly from phenols; the order of impact from the most to the least effected was [970]:

Cladocerans, fish reproduction, algae, bacteria, and rotifers [970].

See also: PAHs as a group entry.

**Br.Car:** Brief Summary of Carcinogenicity/Cancer Information:

Coal tar is among 31 substances classified by the Chief of the Worker Health and Safety Unit of the California Department of Food and Agriculture as having "high carcinogenic or oncogenic potential" (Dr. Keith Maddy, personal communication). There is evidence of carcinogenic effects to animals [609].

The debates on which polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs found in this product to classify as carcinogens, and the details of exactly how to perform both ecological and human risk assessments on the complex mixtures of PAHs typically found at contaminated sites, are likely to continue. There are some clearly wrong ways to go about it, but defining clearly right ways is more difficult. PAHs usually occur in complex mixtures rather than alone. Perhaps the most unambiguous thing that can be said about complex PAH mixtures is that such mixtures are often carcinogenic and possibly phototoxic. One way to approach site specific risk assessments would be to collect the complex mixture of PAHs and other lipophilic contaminants in a semipermeable membrane device (SPMD, also known as a fat bag) [894,895,896], retrieve the contaminant mixture from the SPMD, then test the mixture for carcinogenicity, toxicity, and phototoxicity (James Huckins, National Biological Service, and Roy Irwin, National Park Service, personal communication, 1996).

See also: PAHs as a group entry.

**Br.Dev:** Brief Summary of Developmental, Reproductive, Endocrine, and Genotoxicity Information:

Exposure to coal tar and its components has been linked to embryonic and teratogenic effects, as well as cytotoxicity, in animals [609].

See also: PAHs as a group entry.

**Br.Fate:** Brief Summary of Key Bioconcentration, Fate, Transport, Persistence, Pathway, and Chemical/Physical Information:

Many organic compounds are locked within the structure of coal. When coal is heated in the absence of air, certain volatile gases evolve, including methane and ammonia;

this mixture of gases is called "coal gas." Simultaneously, a relatively viscous liquid condenses near room temperature. The latter material is called "coal tar" and consists of a mixture of many organic compounds, like benzene, toluene, phenol, naphthalene, anthracene, and others [751].

After coal is distilled in coking oven, the coal tar is cooled, collected, and then distilled to yield four fractions and a residue [279]. Light and heavy fractions of coal tar are sometimes differentiated [751].

The first fraction contains benzene, toluene, xylene and other substances [279]. The light coal distillate possesses the lower flash point and thus burns more readily than the heavy coal distillate [751].

The second fraction contains naphthalene and tar acids and bases; the third fraction yields the creosotes and methylnaphthalenes; the fourth fraction distills off quinoline, anthracene, phenanthrene, and carbazole; and the final residue after distillation is called coal-tar pitch [279].

No other summary information found on the fate of this complex mixture; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

See also: PAHs as a group entry.

**Synonyms/Substance Identification:**

CRUDE COAL TAR [609,607]  
ESTAR [609,607]  
LAVATAR [609,607]  
PIX CARBONIS [609,607]  
PIXALBOL [609,607]  
PSORIGEL [609]  
TAR [609,607]  
TAR, COAL [609,607]  
ZETAR [609,607]  
CARBO-CORT [607]  
COAL TAR, AEROSOL [607]  
COAL TAR SOLUTION USP [607]  
LAV [607]  
PICIS CARBONIS [607]  
PIX LITHANTHRACIS [607]  
POLYTAR BATH [607]  
SUPERTAH [607]  
SYNTAR [607]



US Coast Guard CHRIS Database [367]:  
OILS, MISCELLANEOUS: COAL TAR  
CHRIS Code OCT  
Light oil

**Associated Chemicals or Topics (Includes Transformation Products):**

See also individual entries:

Petroleum, General

Various entries on individual PAHs (common components of coal tar).

PAHs as a group entry.

**Metabolism/Metabolites [609]:**

Acridine was detected in the urine of 28 pt receiving coal tar therapy. [Cernikova M et al; Detection of acridine in human urine after topical coal tar treatment; J Chromatogr Biomed Appl 273 (Mar 11): 202-6 (1983)].

Coal Tar and Distillates: Site Assessment-Related Information Provided by Shineldecker (Potential Site-Specific Contaminants that May be Associated with a Property Based on Current or Historical Use of the Property) [490]:

General Types of Associated Materials:

- Coal
- Petroleum fuels
- Pitch
- Solvents

Raw Materials, Intermediate Products, Final Products, and Waste Products Generated During Manufacture and Use:

- Acetone
- Acridine
- Aniline
- Arsenic
- Creosote
- Cresol
- Lead
- Naphthalene
- Phenols
- Polynuclear aromatic hydrocarbons

Other Associated Materials:

- Benzene
- Benzol

- Ethyl benzene
- Toluene
- Xylenes

**Water Data Interpretation, Concentrations and Toxicity (All Water Data Subsections Start with "W."):**

**W.Low** (Water Concentrations Considered Low):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.High** (Water Concentrations Considered High):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Typical** (Water Concentrations Considered Typical):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Concern Levels, Water Quality Criteria, LC50 Values, Water Quality Standards, Screening Levels, Dose/Response Data, and Other Water Benchmarks:**

**W.General** (General Water Quality Standards, Criteria, and Benchmarks Related to Protection of Aquatic Biota in General; Includes Water Concentrations Versus Mixed or General Aquatic Biota):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture. See also: PAHs as a group entry.

**W.Plants** (Water Concentrations vs. Plants):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Invertebrates** (Water Concentrations vs. Invertebrates):

No information found; see Chem.Detail section for

compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Fish** (Water Concentrations vs. Fish):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Wildlife** (Water Concentrations vs. Wildlife or Domestic Animals):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Human** (Drinking Water and Other Human Concern Levels):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**W.Misc.** (Other Non-concentration Water Information):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sediment Data Interpretation, Concentrations and Toxicity** (All Sediment Data Subsections Start with "Sed."):

**Sed.Low** (Sediment Concentrations Considered Low):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.High** (Sediment Concentrations Considered High):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Typical** (Sediment Concentrations Considered Typical):

No information found; see Chem.Detail section for

compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Concern Levels, Sediment Quality Criteria, LC50 Values, Sediment Quality Standards, Screening Levels, Dose/Response Data and Other Sediment Benchmarks:**

**Sed.General** (General Sediment Quality Standards, Criteria, and Benchmarks Related to Protection of Aquatic Biota in General; Includes Sediment Concentrations Versus Mixed or General Aquatic Biota):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture. See also: PAHs as a group entry.

**Sed.Plants** (Sediment Concentrations vs. Plants):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Invertebrates** (Sediment Concentrations vs. Invertebrates):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Fish** (Sediment Concentrations vs. Fish):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Wildlife** (Sediment Concentrations vs. Wildlife or Domestic Animals):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Sed.Human** (Sediment Concentrations vs. Human):

No information found; see Chem.Detail section for compounds in this product, then see individual

compound entries for summaries of information on individual components of this mixture.

**Sed.Misc.** (Other Non-concentration Sediment Information):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil** Data Interpretation, Concentrations and Toxicity (All Soil Data Subsections Start with "Soil."):

**Soil.Low** (Soil Concentrations Considered Low):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil.High** (Soil Concentrations Considered High):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil.Typical** (Soil Concentrations Considered Typical)

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil.Concern Levels**, Soil Quality Criteria, LC50 Values, Soil Quality Standards, Screening Levels, Dose/Response Data and Other Soil Benchmarks:

**Soil.General** (General Soil Quality Standards, Criteria, and Benchmarks Related to Protection of Soil-dwelling Biota in General; Includes Soil Concentrations Versus Mixed or General Soil-dwelling Biota):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture. See also: PAHs as a group entry.

**Soil.Plants** (Soil Concentrations vs. Plants):

No information found; see Chem.Detail section for compounds in this product, then see individual

compound entries for summaries of information on individual components of this mixture.

**Soil.Invertebrates** (Soil Concentrations vs. Invertebrates):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil.Wildlife** (Soil Concentrations vs. Wildlife or Domestic Animals):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Soil.Human** (Soil Concentrations vs. Human):

No information found on this complex and variable mixture. See Chem.Detail section for chemicals found in this product, then look up information on each hazardous compound. Some individual compounds found in petroleum products have low-concentration human health benchmarks for soil (see individual entries for these compounds).

**Soil.Misc.** (Other Non-concentration Soil Information):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tissue and Food Concentrations** (All Tissue Data Interpretation Subsections Start with "Tis."):

**Tis.Plants:**

A) As Food: Concentrations or Doses of Concern to Living Things Which Eat Plants:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

B) Body Burden Residues in Plants: Typical, Elevated, or of Concern Related to the Well-being of the Organism Itself:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tis.Invertebrates:**

A) As Food: Concentrations or Doses of Concern to Living Things Which Eat Invertebrates:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

B) Concentrations or Doses of Concern in Food Items Eaten by Invertebrates:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

C) Body Burden Residues in Invertebrates: Typical, Elevated, or of Concern Related to the Well-being of the Organism Itself:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tis.Fish:**

A) As Food: Concentrations or Doses of Concern to Living Things Which Eat Fish (Includes FDA Action Levels for Fish and Similar Benchmark Levels From Other Countries):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

B) Concentrations or Doses of Concern in Food Items Eaten by Fish:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

C) Body Burden Residues in Fish: Typical, Elevated, or of Concern Related to the Well-being of the Organism Itself:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tis.Wildlife:** Terrestrial and Aquatic Wildlife, Domestic Animals and all Birds Whether Aquatic or not:

A) As Food: Concentrations or Doses of Concern to Living Things Which Eat Wildlife, Domestic Animals, or Birds:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

B) Concentrations or Doses of Concern in Food Items Eaten by Wildlife, Birds, or Domestic Animals (Includes LD50 Values Which do not Fit Well into Other Categories, Includes Oral Doses Administered in Laboratory Experiments):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

C) Body Burden Residues in Wildlife, Birds, or Domestic Animals: Typical, Elevated, or of Concern Related to the Well-being of the Organism Itself:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tis.Human:**

A) Typical Concentrations in Human Food Survey Items:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

B) Concentrations or Doses of Concern in Food Items Eaten by Humans (Includes Allowable Tolerances in Human Food, FDA, State and Standards of Other Countries):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on



individual components of this mixture.

C) Body Burden Residues in Humans: Typical, Elevated, or of Concern Related to the Well-being of Humans:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Tis.Misc.** (Other Tissue Information):

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Bio.Detail:** Detailed Information on Bioconcentration, Biomagnification, or Bioavailability:

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Interactions:**

No information found; see Chem.Detail section for compounds in this product, then see individual compound entries for summaries of information on individual components of this mixture.

**Uses/Sources:**

Coal tar is a viscous black liquid derived from the destructive distillation of coal, used in many dyes, drugs, and organic chemicals and for waterproofing, paints, roofing, and insulation materials [492].

Major Uses [609]:

Raw material for plastics, solvents, dyes, drugs; crude or refined product or fractions used for waterproofing, paints, pipecoating, roads, roofing, insulation, as pesticides & sealants [Hawley, G.G. The Condensed Chemical Dictionary. 10th ed. New York: Van Nostrand Reinhold Co., 1981. 258].

Distillation to produce volatile fractions & pitch; direct burning as a fuel; component of products for treatment of psoriasis; alcohol denaturant; binder & filler in surface coatings; modifier in epoxy resin surface coatings.

Artificial Sources [609]:

The chemical and physical nature emitted from spreading applications of coal tar were characterized. [Hittle DC, Stukel JJ; Particle size distribution and chemical composition of coal-tar fumes; Am Ind Hyg Assoc J 37 (4): 199-204 (1976)]

**Forms/Preparations/Formulations:**

Formulations/Preparations [609]:

Coal tar (liquor carbonis detergens), usp: generic: soln 20%. This soln contains emulsifier, polysorbate 80 (tween 80)... [American Medical Association, AMA Department of Drugs. AMA Drug Evaluations. 5th ed. Chicago: American Medical Association, 1983. 1365].

Coal tar, usp is used in 1-5% strengths in creams, gels, lotions, ointments, pastes, & shampoos & in 5, 20, or 30% concn in topical soln, emulsions, or suspensions, & as 50% bath emulsion. Coal tar is also combined with salicylic acid & sulfur in antifungal prepn. [Gilman, A. G., L. S. Goodman, and A. Gilman. (eds.). Goodman and Gilman's The Pharmacological Basis of Therapeutics. 6th ed. New York: Macmillan Publishing Co., Inc. 1980. 970].

Grades: crude, refined; usp. [Hawley, G.G. The Condensed Chemical Dictionary. 10th ed. New York: Van Nostrand Reinhold Co., 1981. 258].

**Chem.Detail:** Detailed Information on Chemical/Physical Properties:

Caution: Every individual petroleum product has a unique "fingerprint," or distinct set of constituents most commonly identified by a gas chromatograph analysis. Due to the varying properties of the same general category of a petroleum product (each source and weathering stage of Coal tar has a unique gas chromatograph "fingerprint"), careful assessment of the toxicity, specific gravity, and other physical characteristics of each individual oil must be taken into consideration to determine the exact effects of the product on the environment. Therefore, the below comments on Coal tar are to be considered as representative, but not absolute values typical of every batch of the product with the same name.

Since PAHs are important hazardous components of this product, risk assessments should include analyses of PAHs and alkyl PAHs utilizing the NOAA protocol expanded scan [828] or other rigorous GC/MS/SIM methods. See also: PAHs as a group entry.

Coal tar contains a great variety of mononuclear and polynuclear aromatic hydrocarbons (PAHs) [749]. Among the hazardous substances in coal tar are benzene, toluene, phenol, naphthalene, anthracene, and others [751]. It also contains benzo(a)pyrene, a strong carcinogen [366].

An appropriate laboratory technique for identifying PAHs and alkyl PAHs is the NOAA expanded scan [828].

Density/Specific Gravity [609]:

1.18 TO 1.23 [Hawley, G.G. The Condensed Chemical Dictionary. 10th ed. New York: Van Nostrand Reinhold Co., 1981. 258 [609]

Specific gravity:(est.) 0.90 at 20 degrees C (liquid) [367].

Solubilities [609]:

Small portion of coal tar dissolves in water; all or almost all dissolves in benzene or nitrobenzene; partly dissolves in alc, ether, chloroform, methanol, carbon disulfide, acetone, petroleum ether, sodium hydroxide soln [The Merck Index. 10th ed. Rahway, New Jersey: Merck Co., Inc., 1983. 345].

Partially dissolved by hexane [Osol, A. (ed.). Remington's Pharmaceutical Sciences. 16th ed. Easton, Pennsylvania: Mack Publishing Co., 1980. 724].

Pixalbol: practically insol in water; sol in 20 parts alc; miscible with abs alc, acetone, oil & fats, petrolatum [The Merck Index. 10th ed. Rahway, New Jersey: Merck Co., Inc., 1983. 345].

Color/Form [609]:

Almost black, thick liq or semisolid [The Merck Index. 10th ed. Rahway, New Jersey: Merck Co., Inc., 1983. 345].

Brownish, viscous substance [International Labour Office. Encyclopedia of Occupational Health and Safety. Vols. I&II. Geneva, Switzerland: International Labour Office, 1983. 2147].

Odor [609]:

Characteristic naphthalene-like odor [Osol, A. (ed.). Remington's Pharmaceutical Sciences. 16th ed. Easton, Pennsylvania: Mack Publishing Co., 1980. 724].

Taste [609]:

Sharp burning taste [Osol, A. (ed.). Remington's Pharmaceutical Sciences. 16th ed. Easton, Pennsylvania: Mack Publishing Co., 1980. 725].

Surface Tension [609]:

25 dynes/cm= 0.025 N/m @ 20 deg C [U.S. Coast Guard, Department of Transportation. CHRIS - Hazardous Chemical Data. Manual Two. Washington, DC: U.S. Government Printing Office, Oct., 1978.

Other Chemical/Physical Properties [609]:

Heavier than water; on ignition it burns with reddish, luminous, & very sooty flame [Osol, A. (ed.). Remington's Pharmaceutical Sciences. 16th ed. Easton, Pennsylvania: Mack Publishing Co., 1980. 724].

Pixalbol: A Light-yellow, thin, oily liquid; Practically insol in water; sol in 20 parts alcohol; miscible with absolute alcohol, acetone, petroleum, oils & fats [The Merck Index. 10th ed. Rahway, New Jersey: Merck Co., Inc., 1983. 345].

Information from CHRIS [367]:

PHYSICAL AND CHEMICAL PROPERTIES

Physical state at 15 degrees C. and 1 ATM:Liquid  
Molecular weight:Not pertinent  
Boiling point at 1 ATM:223-333 degrees F = 106-167 degrees C = 379-440 degrees K  
Freezing point:Not pertinent  
Critical temperature:Not pertinent  
Critical pressure:Not pertinent  
Liquid surface tension:(est.) 25 dynes/cm = 0.025 N/m at 20 degrees C  
Liquid water interfacial tension:Data not available  
Vapor (gas) specific gravity:Not pertinent  
Ratio of specific heats of vapor (gas):(est.) 1.071  
Latent heat of vaporization:(est.) 107 Btu/lb = 59.8 cal/g = 2.5 X 10<sup>(5)</sup> J/kg  
Heat of combustion:-17,440 Btu/lb = -9,690 cal/g = -405.7 X 10<sup>(5)</sup> J/kg  
Heat of decomposition:Not pertinent  
Heat of solution:Not pertinent  
Heat of polymerization:Not pertinent  
Heat of fusion:Data not available  
Limiting value:Data not available  
REID vapor pressure:Data not available  
Characteristics:  
Liquid Colorless to yellow Pleasant odor  
Floats on water. Flammable, irritating vapor is produced.

OBSERVABLE CHARACTERISTICS

Physical state:Liquid  
Color:Colorless to yellow  
Odor:Aromatic; like benzene; pleasant

**Fate.Detail:** Detailed Information on Fate, Transport, Persistence, and/or Pathways:

Absorption, Distribution and Excretion [609]:

Deepest penetration of 5% coal tar prepn into rabbit skin was with wool fat base, followed by white petrolatum, o/w base, absorption base and hydrogel. Highest retention 10 days after last admin was obtained with white petrolatum. [RITSCHEL WA ET AL; BIOPHARMACEUTICAL EVALUATION OF TOPICAL TAR PREPARATIONS; SCI PHARM 43 (MAR 31): 11-21 (1975)].

**Laboratory and/or Field Analyses:**

Coal tars contain many of the same PAHs and other aromatics as do many petroleum products, so the lab methods used are similar. However, recent (1991) studies have indicated that EPA approved methods used for oil spill assessments (including total petroleum hydrocarbons method 418.1, semivolatile priority pollutant organics methods 625 and 8270, and volatile organic priority pollutant methods 602, 1624, and 8240) are all inadequate for generating scientifically defensible information for Natural Resource Damage Assessments [468]. These general organic chemical methods are deficient in chemical selectivity (types of constituents analyzed) and sensitivity (detection limits); the deficiencies in these two areas lead to an inability to interpret the environmental significance of the data in a scientifically defensible manner [468]. See also: PAHs as a group entry.

Light and heavy fractions of coal tar are sometimes differentiated [751]. The lab analyses most appropriate for measuring different types of coal tar contamination depend upon the type of product involved and the reason for measuring the contamination. The farther one progresses from lighter towards heavier coal tars, the greater the percentage of PAHs and other semi-volatiles (many of which are not so immediately toxic as the volatiles but which can result in long term/chronic impacts). Heavier coal tars thus need to be analyzed for the semi volatile compounds which typically pose the greatest long term risk, PAHs and (especially) alkylated PAHs.

Due to the presence of both light aromatics and heavier PAHs in some coal tars, we recommend the following decision tree:

Decision Tree (dichotomous key) for selection of lab methods for measuring contamination from midrange to heavy products:

- 1a. Your main concern is biological effects .....2
- 1b. Your main concern is cleanup or remediation  
but no ecological or human resources are at risk.....3
- 2a. The resource at risk is primarily humans via a drinking water pathway, either the contamination of groundwater used for drinking water, or the fresh\* or continuing contamination of surface waters used as drinking water, or the risk is primarily to aquatic species in confined\*\* surface waters from a fresh\* spill, or the risk is to surface waters re-emerging from contaminated groundwater resources whether the spill is fresh\* or not; the medium and/or pathway of concern is water

rather than sediments, soil, or tissues. Note: although heavy products have a lower percentage of BTEX and other relatively soluble compounds which typically threaten drinking water, ground water, or water column organisms, some heavy products do contain some of these water soluble compounds, so they cannot be ignored.....4

2b. The resource at risk is something else.....5

3a. The spilled substance is a fresh\* product of known composition: If required to do so by a regulatory authority, perform whichever Total Petroleum Hydrocarbon (TPH) analysis specified by the regulator. However, keep in mind that due to its numerous limitations, the use of the common EPA method 418.1 for Total Petroleum Hydrocarbons is not recommended as a stand-alone method unless the results can first be consistently correlated (over time, as the product ages) with the better NOAA protocol expanded scan\*\*\* for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs. If not required to perform an EPA method 418.1-based analysis for TPH, instead perform a Gas Chromatography/Flame Ionization Detection (GC/FID) analysis for TPH using the spilled substance as a calibration standard. GC/FID methods can be sufficient for screening purposes when the coal tar contamination is fresh\*, unweathered coal tar and when one is fairly sure of the source [657]. If diesel 1D was spilled, perform TPH-D (1D) using California LUFT manual methods (typically a modified EPA method 8015) [465] or a locally available GC/FID method of equal utility for the product spilled. However, no matter which TPH method is used, whether based on various GC/FID or EPA method 418.1 protocols, the investigator should keep in mind that the effectiveness of the method typically changes as coal tar ages, that false positives or false negatives are possible, and that the better Gas Chromatography-Mass Spectrometry-Selected Ion Mode (GC/MS/SIM) scans (such as the NOAA expanded scan\*\*\*) should probably be performed at the end of remediation to be sure that the contamination has truly been cleaned up. Another option for fresh coal tar: in cases where an inexpensive screening scan is desired, consider using an HPLC/Fluorescence scan method for sediment or bile metabolite samples. Such scans are available from laboratories at Texas A. and M., Arthur D. Little, and the NOAA lab in Seattle. This scan is not much more expensive, and less prone to false negatives and various other problems than some of the more common screening methods (TPH-EPA 418.1 and Oil and Grease). Screening measures the total fluorescence of coal tar components while GC/MS measures individual aromatic compounds [521]. Thus, HPLC/fluorescence screening allowed detecting lower concentrations of petroleum-related aromatic compounds in samples contaminated by Prudhoe Bay Crude Oil than did analysis by GC/MS [521].

3b. The spilled product is not fresh\* or the contamination is of unknown or mixed composition.....6

4. Analyze for Benzene, Toluene, Ethyl Benzene, and Toluene (BTEX) compounds in water as part of a broader scan of volatiles using EPA GC/MS method 8240. The standard EPA GC/MS method 8240 protocol will be sufficient for some applications, but the standard EPA method 8240 (and especially the less rigorous EPA BTEX methods such as method 8020 for soil and method 602 for water) are all inadequate for generating scientifically defensible information for Natural Resource Damage Assessments [468]. The standard EPA methods are also inadequate for risk assessment purposes. Thus, when collecting information for possible use in a Natural Resource Damage Assessment or risk assessment, it is best to ask the lab to analyze for BTEX compounds and other volatile coal tar compounds using a modified EPA GC/MS method 8240 method using the lowest possible Selected Ion Mode detection limits and increasing the analyte list to include as many alkyl BTEX compounds as possible. Also analyze surface or (if applicable) ground water samples for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs using the NOAA protocol expanded scan\*\*\* modified for water samples using methylene chloride extraction. If the contaminated water is groundwater, before the groundwater is determined to be remediated, also analyze some contaminated sub-surface soils in contact with the groundwater for BTEX compounds (EPA GC/MS method 8240), and PAHs (NOAA protocol expanded scan\*\*\*). The magnitude of any residual soil contamination will provide insight about the likelihood of recontamination of groundwater resources through equilibria partitioning mechanisms moving contamination from soil to water.
  
- 5a. The medium of concern is sediments or soils.....6
- 5b. The medium of concern is biological tissues.....7
  
6. Perform the NOAA protocol expanded scan\*\*\* for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs. If there is any reason to suspect fresh\* or continuing contamination of soils or sediments with lighter volatile compounds, also perform EPA GC/MS method 8240 using the lowest possible Selected Ion Mode (SIM) detection limits and increasing the analyte list to include as many alkyl Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX) compounds as possible.
  
- 7a. The problem is direct coating (oiling) of wildlife or plants with spilled coal tar product.....8
- 7b. The problem is something else.....9
  
8. Perform NOAA protocol expanded scan\*\*\* for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs and/or GC/FID fingerprinting of the coating oil only if necessary to identify the source or exact oil. If the source is known and no confirmation lab studies are necessary: dispense with additional chemical laboratory analyses and instead document

direct effects of coating: lethality, blinding, decreased reproduction from eggshell coating, etc., and begin cleaning activities if deemed potentially productive after consultations with the Fish and Wildlife Agencies.

- 9a. The concern is for impacts on water column organisms (such as fish or plankton).....10
- 9b. The concern is for something else (including benthic organisms).....11
- 10. If exposure to fish is suspected, an HPLC/Fluorescence scan for polycyclic aromatic hydrocarbon (PAH) metabolites in bile may be performed to confirm exposure [844]. The HPLC/fluorescence scan looks for the presence of metabolites of PAHs: naphthalene, phenanthrene, and benzo[a]pyrene. The technique does not identify or quantify actual PAH compounds, but subsequent gas chromatography analyses can be done to confirm the initial findings. Even the semi-quantitative Total Scanning Fluorescence (TSF) done inexpensively by labs such as GERG are a better measure of PAH contamination than GC/FID, which measures less persistent and less hazardous aliphatics. For bottom-dwelling fish such as flounders or catfish, also analyze the bottom sediments (see Step 6 above). Fish which spend most of their time free-swimming above the bottom in the water column can often avoid toxicity from toxic aromatic compounds in the water column, but if fish are expiring in a confined\*\* habitat (small pond, etc.), EPA GC/MS method 8240 and the NOAA protocol expanded scan\*\*\* for PAHs could be performed to see if Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX), naphthalene, and other potentially toxic compounds are above known acute toxicity benchmark concentrations. Zooplankton populations impacted by aromatics usually recover fairly quickly unless they are impacted in very confined\*\* or shallow environments [835] and the above BTEX and PAH water methods are often recommended rather than direct analyses of zooplankton tissues.
- 11a. The concern is for benthic invertebrates: analyze invertebrate whole-body tissue samples and surrounding sediment samples for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs using the NOAA protocol expanded scan\*\*\*. If the spill is fresh\* or the source continuous, risk assessment needs may also require that the sediments which form the habitat for benthic invertebrates be analyzed for Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX) and other volatile compounds using EPA GC/MS method 8240 or modified EPA method 8240 in the Selected Ion Mode (SIM). Bivalve invertebrates such as clams and mussels do not break down PAHs as well or as quickly as do fish or many wildlife species. They are also less mobile. Thus, bivalve tissues are more often directly analyzed for PAH residues than are the tissues of fish or wildlife.
- 11b. The concern is for plants or for vertebrate wildlife including



birds, mammals, reptiles, and amphibians: polycyclic aromatic hydrocarbons (PAHs) and other coal tar hydrocarbons break down fairly rapidly in many wildlife groups and tissues are not usually analyzed directly. Instead direct effects are investigated and water, soil, sediment, and food items encountered by wildlife are usually analyzed for PAHs and alkyl PAHs using the NOAA protocol expanded scan\*\*\*. If the spill is fresh\* or the source continuous, risk assessment needs may also require that these habitat media also be analyzed for Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX) and other volatile compounds using EPA GC/MS method 8240 or modified EPA method 8240 in the Selected Ion Mode (SIM). Less is known about plant effects. However, the same methods recommended above for the analyses of water (Step 4 above) and for sediments or soils (Step 6 above) are usually also recommended for these same media in plant or wildlife habitats. If wildlife or plants are covered with oil, see also Step 8 (above) regarding oiling issues.

\* Discussion of the significance of the word "fresh": The word "fresh" cannot be universally defined because coal tar breaks down faster in some environments than in others. In a hot, windy, sunny, oil-microbe-rich, environment in the tropics, some of the lighter and more volatile compounds (such as the Benzene, Toluene, Ethyl Benzene, and Xylene compounds) would be expected to disappear faster by evaporation into the environment and by biodegradation than in a cold, no-wind, cloudy, oil-microbe-poor environment in the arctic. In certain habitats, BTEX and other relatively water soluble compounds will tend to move to groundwater and/or subsurface soils (where degradation rates are typically slower than in a sunny well aerated surface environment). Thus, the judgement about whether or not aromatic contamination would be considered "fresh" is a professional judgement based on a continuum of possible scenarios. The closer in time to the original spill of non-degraded product, the greater degree the source is continuous rather than the result of a one-time event, and the more factors are present which would retard aromatic evaporation or breakdown (cold, no-wind, cloudy, oil-microbe-poor conditions, etc.) the more likely it would be that in the professional judgement experts the product would be considered "fresh." In other words, the degree of freshness is a continuum which depends on the specific product spilled and the specific habitat impacted. Except for groundwater resources (where the breakdown can be much slower), the fresher the middle distillate product contamination is, the more one has to be concerned about potential impacts of BTEX compounds, and other lighter and more volatile compounds.

To assist the reader in making decisions based on the continuum of possible degrees of freshness, the following generalizations are provided: Some of the lightest middle distillates are moderately volatile and soluble and up to two-thirds of the spill amount could disappear from surface waters after a few days [771,835]. Typically the volatile fractions disappear mostly by evaporating into the atmosphere. However, in some cases, certain water soluble

fractions of coal tar including Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX) compounds move down into groundwater. BTEX compounds are included in the more volatile and water soluble fractions, and BTEX compounds as well as the lighter alkanes are broken down more quickly by microbes than heavier semi-volatiles such as alkyl PAHs and some of the heavier and more complex aliphatic compounds. Thus after a week, or in some cases, after a few days, there is less reason to analyze surface waters for BTEX or other volatile compounds, and such analyses should be reserved more for potentially contaminated groundwaters. In the same manner, as the product ages, there is typically less reason to analyze for alkanes using GC/FID techniques or TPH using EPA 418.1 methods, and more reason to analyze for the more persistent alkyl PAHs using the NOAA protocol expanded scan\*\*\*.

\*\* Discussion of the significance of the word "confined": Like the word "fresh" the word "confined" is difficult to define precisely as there is a continuum of various degrees to which a habitat would be considered "confined" versus "open." However, if one is concerned about the well-being of ecological resources such as fish which spend most of their time swimming freely above the bottom, it makes more sense to spend a smaller proportion of analytical funding for water column and surface water analyses of Benzene, Toluene, Ethyl Benzene, and Xylene (BTEX) and other volatile or acutely toxic compounds if the spill is in open and/or deep waters rather than shallow or "confined" waters. This is because much of the contamination tends to stay with a surface slick or becomes tied up in subsurface tar balls. The coal tar compounds which do pass through the water column often tend to do so in small concentrations and/or for short periods of time, and fish and other pelagic or generally mobile species can often swim away to avoid impacts from spilled coal tar in "open waters." Thus in many large spills in open or deep waters, it has often been difficult or impossible to attribute significant impacts to fish or other pelagic or strong swimming mobile species in open waters. Lethality has most often been associated with heavy exposure of juvenile fish to large amounts of coal tar products moving rapidly into shallow or confined waters [835]. Different fish species vary in their sensitivity to aromatics [835]. However, the bottom line is that in past ecological assessments of spills, often too much money has been spent on water column analyses in open water settings, when the majority of significant impacts tended to be concentrated in other habitats, such as benthic, shoreline, and surface microlayer habitats.

\*\*\* The expanded scan protocols for polycyclic aromatic hydrocarbons (PAHs) and alkyl PAHs have been published by NOAA [828].

Additional Information:

Since coal tar has many of the same PAHs as do petroleum products, the following discussion is relevant:

Although EPA method 418.1: Petroleum Hydrocarbons expressed as Total Petroleum Hydrocarbons (TPH), is recommended by many State agencies, some consulting firms, and some laboratories for certain regulatory and screening applications (often leaking underground storage tanks), this method is not well suited to coal tar contamination or to the more persistent hazardous constituents in this product. Low values tend to give the mistaken impression that a site is clean when it really isn't (a false negative). For example, a field test of bioremediation of soils contaminated with Bunker C (a heavy fuel) at a refinery in Beaumont, Texas, utilized oil and grease data, which (although the data was quite variable) seemed to indicate bioremediation was taking place [728]. A comparison of the oil and grease data at this site with TPH data at this site suggested the same thing, that the data was quite variable but if anything, the oil was being slowly being cleaned up by bioremediation (Bruce Herbert, Texas A. and M., Department of Geology, personal communication, 1995). However, a later study of the same site utilizing the expanded scan for PAHs [828] (a modified EPA 8270 including alkyl homologues and lower detection limits), indicated that very little bioremediation of hazardous alkyl PAHs and multi-ring PAHs was actually taking place [727]. Thus, utilizing either oil and grease or TPH analyses would tend to lead one to the faulty conclusion that the harmful compounds were being naturally cleaned up at an acceptable rate. This is partly because the TPH and oil and grease methods tend to favor the lighter and less alkylated PAHs, whereas many of the carcinogenic and longer lasting PAHs are the heavier multi-ringed and alkylated compounds.

See also:

Laboratory and/or Field Analyses section in Oil Spills entry for information on biological indicators of exposure to PAHs and other aromatics.

PAHs as a group entry.